## MATE 201

Fall 2019

## Homework \# 3

Due date: November 4 ${ }^{\text {th }}, 2019$
(lecture time)

## No late submissions!

Group submission (up to 4 students per group) is allowed.

Your homework submission should have a cover page which contains the following information;
your name, student number, course name, homework number and date of submission.

## Question 3:

Compute the atomic packing factor for the rock salt crystal structure in which $\mathrm{r}_{\mathrm{C}} / \mathrm{r}_{\mathrm{A}}=\mathbf{0 . 4 1 4}$

## Question 4:

Calculate the theoretical density of FeO , given that it has the rock salt crystal structure.

## Question 1:

What point defects are possible for MgO as an impurity in $\mathrm{Al}_{2} \mathrm{O}_{3}$ ? How many $\mathrm{Mg}^{2+}$ ions must be added to form each of these defects?

## Question 2:

Based on the ionic charge and ionic radii, predict crystal structures for the following materials: (a) NiS, (b) KBr , and (c) CsBr.

## Question 5:

Calculate the number of Frenkel defects per cubic centimeter in silver chloride at $300^{\circ} \mathrm{C}$. The energy for defect formation is 1.1 eV , while the density for AgCI is $5.50 \mathrm{~g} / \mathrm{cm}^{3}$ at $300^{\circ} \mathrm{C}$.

$$
N_{f r}=N \exp \left(-\frac{Q_{f r}}{2 k T}\right)
$$

## Question 6:

Calculate the number of Frenkel defects per cubic centimeter in silver chloride at $300^{\circ} \mathrm{C}$. The energy for defect formation is 1.1 eV , while the density for AgCl is $5.50 \mathrm{~g} / \mathrm{cm}^{3}$ at $300^{\circ} \mathrm{C}$.

## Question 7:

The modulus of elasticity for spinel $\left(\mathrm{MgAl}_{2} \mathrm{O}_{4}\right)$ having 4 vol\% porosity is 245.3 GPa .
a) Compute the modulus of elasticity for the nonporous material.
b) Compute the modulus of elasticity for 9 vol\% porosity.

## Question 8:

a) Compute the repeat unit molecular weight of polypropylene.
b) Compute the number-average molecular weight for a polypropylene for which the degree of polymerization is 21000 .

## Question 9:

Molecular weight data for some polymer are tabulated here. Compute
(a) The number average molecular weight, and
(b) The weight-average molecular weight
(c) If it is known that this material's degree of polymerization is 760 , which polymer is this?

| Molecular Weight |  |  |
| :---: | :---: | :---: |
| Range (g/mol) | xi | wi |
| $8,000-20,000$ | 0.05 | 0.02 |
| $20,000-32,000$ | 0.15 | 0.08 |
| $32,000-44,000$ | 0.21 | 0.17 |
| $44,000-56,000$ | 0.28 | 0.29 |
| $56,000-68,000$ | 0.18 | 0.23 |
| $68,000-80,000$ | 0.10 | 0.16 |
| $80,000-92,000$ | 0.03 | 0.05 |

